Brønsted Acid-Catalyzed, Enantioselective, Vinylogous Mannich Reaction of Vinylketene Silyl *N,O*-Acetals

David S. Giera, Marcel Sickert, and Christoph Schneider*

Institut für Organische Chemie, Universität Leipzig, Johannisallee 29, 04103 Leipzig, Germany

schneider@chemie.uni-leipzig.de

Received July 29, 2008

ORGANIC LETTERS 2008 Vol. 10, No. 19 4259-4262

ABSTRACT



Vinylketene silyl *N*,*O*-acetals undergo chiral phosphoric acid-catalyzed, vinylogous Mukaiyama–Mannich reactions with imines and afford δ -amino- α , β -unsaturated amides in typically good yields, complete γ -regioselectivity, and up to 92% ee with catalyst loadings of as low as 1 mol %. The Mannich products can be readily manipulated to furnish valuable synthetic intermediates.

Vinylogous carbon–carbon bond-forming reactions of metal dienolates are very important transformations in organic synthesis and deliver highly functionalized products of great utility. Asymmetric vinylogous aldol reactions furnishing δ -hydroxy α,β -unsaturated carboxylic acid derivatives have been widely studied in the context of the total synthesis of polyketide-derived natural products employing chiral auxiliaries as well as chiral catalysts.¹ The development of catalytic, enantioselective vinylogous Mannich reactions, however, lags behind, although this strategy holds great potential for natural product synthesis, in particular, alkaloid

synthesis.² Thus far, only a few protocols have been devised that furnish enantiomerically highly enriched products, and these methods are frequently restricted to special substrate patterns. Following precedence established by Martin,³ Hoveyda and Snapper et al. developed silver-catalyzed, vinylogous Mannich reactions of 2-silyloxy furans giving rise to highly enantiomerically enriched butenolides.⁴ A Brønsted acid-catalyzed approach to chiral γ -butenolides was described by Akiyama et al.⁵ Chen and co-workers recently established the first direct vinylogous asymmetric Mannich reaction of α , α -dicyano alkenes and Boc-protected imines with a bifunctional thiourea catalyst leading to the corresponding products in high yields and excellent enantiose-

10.1021/ol8017374 CCC: 40.75 $$\odot$$ 2008 American Chemical Society Published on Web 08/28/2008

For reviews, see: (a) Denmark, S. E.; Heemstra, J. R.; Beutner, G. L. Angew. Chem., Int. Ed. 2005, 44, 4682. (b) Kalesse, M. Top. Curr. Chem. 2005, 244, 43. (c) Casiraghi, G.; Zanardi, F.; Appendino, G.; Rassu, G. Chem. Rev. 2000, 100, 1929. For select recent work, see: (d) Evans, D. A.; Burch, J. D.; Essa, H.; Jaeschke, G. Tetrahedron 2008, 64, 4671. (e) Curti, C.; Sartori, A.; Battistini, L.; Rassu, G.; Burreddu, P.; Zanardi, F.; Casiraghi, G. J. Org. Chem. 2008, 73, 5446. (f) Simsek, S.; Horzella, M.; Kalesse, M. Org. Lett. 2007, 9, 5637. (g) Heumann, L. V.; Keck, G. E. Org. Lett. 2007, 9, 4275. (h) Denmark, S. E.; Heemstra, J. R. J. Org. Chem. 2007, 72, 5668.
 (i) Shirokawa, S.; Shinoyama, M.; Ooi, I.; Hosokawa, S.; Nakazaki, A.; Kobayashi, S. Org. Lett. 2007, 9, 849.

⁽²⁾ For reviews, see: (a) Martin, S. F. Acc. Chem. Res. 2002, 35, 895.
(b) Bur, S. K.; Martin, S. F. Tetrahedron 2001, 57, 3221.

⁽³⁾ Martin, S. F.; Lopez, O. D. *Tetrahedron Lett.* **1999**, *40*, 8949.

⁽⁴⁾ Carswell, E. L.; Snapper, M. L.; Hoveyda, A. H. Angew. Chem., Int. Ed. **2006**, 45, 7230.

⁽⁵⁾ Akiyama, T.; Honma, Y.; Itoh, J.; Fuchibe, K. Adv. Synth.Catal. 2008, 350, 399.

lectivities as well as complete regioselectivity.⁶ Jørgensen and co-workers catalyzed the same reaction successfully under phase-transfer conditions with a chiral pyrrolidinium salt in up to 95% ee.⁷ Very recently, Shibasaki and co-workers disclosed a direct vinylogous Mannich reaction of γ -butenolides catalyzed through a chiral lanthanum complex.⁸

We have previously reported the first catalytic, enantioselective, vinylogous Mukaiyama–Mannich reacion of vinylketene silyl *O*,*O*-acetal **2** derived from an α , β -unsaturated ester and imines **1**.⁹ δ -Amino- α , β -unsaturated esters **3** were obtained in typically high yields, complete γ -regioselectivity, and good to very good enantioselectivities (Scheme 1). A



BINOL-based phosphoric acid **4a** with 3,3'-mesityl groups was employed as chiral Brønsted acid catalyst which led to protonation of the imines and formed chiral contact ion pairs in situ. This general strategy had been established independently by the groups of Akiyama and Terada^{10,11} and was subsequently applied to a broad range of mainly imine addition reactions by various groups.^{12–16}

We have now extended this process to vinylogous Mukaiyama–Mannich reactions of vinylketene silyl *N*,*O*-

acetals 5 and imines. In a first step, we investigated various dienolates 5 differing in the amine component in reactions with imine **1a** (Table 1). As reaction conditions we employed our previously optimized protocol using 3 equiv of the respective vinylketene silvl N.O-acetal 5, an alcoholic solvent system containing equal amounts of 2-propanol, 2-methyl-2-propanol, and 2-methyl-2-butanol with 1 equiv of water and 10 mol % of phosphoric acid 4a. Whereas the N.Ndimethyl-, N,N-diisopropylamide-, and pyrrolidide-based silyl dienolates 5a-c, respectively, did not perform to our expectations (entries 1-3), the piperidide- and morpholidederived vinylketene silyl N,O-acetals 5d and 5e proved to be more suitable substrates and delivered the corresponding vinylogous Mannich products 9a and 10a in good yields and up to 78% ee within short reaction times of just 30 min at -30 °C (entries 4 and 5).

Changing the Brønsted acid catalyst from our previously optimized 3,3'-bismesityl-substituted phosphoric acid **4a** to the 3,3'-bistriphenylsilyl-substituted analogue **4b**^{12c} had a dramatic effect on the enantioselectivity of the reaction. The vinylogous Mannich product **9a** was now obtained in 91% yield and 90% ee whereas **10a** was isolated in 80% yield and 67% ee (entries 6 and 7). In both cases, the opposite enantiomers were formed using the identical *R*-BINOL-enantiomer as catalyst backbone. The same phenomenon had been observed earlier by You et al. in the asymmetric transfer hydrogenation of β , γ -alkynyl α -imino esters.¹⁷ On the basis of these results, we selected the piperidide-based vinylketene silyl *N*,*O*-acetal **5d** and phosphoric acid **4b** to study this reaction further.

The catalyst loading could be lowered to just 1 mol % by simply extending the reaction time with no detrimental effects on yield or enantioselectivity (Table 2). Thus, reaction of imine **1b** and vinylketene silyl *N*,*O*-acetal **5d** with 1 mol % of Brønsted acid **4b** furnished vinylogous Mannich product **9b** with 99% yield and 90% ee after 7 d at -30 °C even slightly exceeding the results obtained with higher catalyst loadings (compare entries 1 and 4).

Table 3 shows that the optimized protocol is broadly applicable to aromatic and heteroaromatic aldimines 1b-o which were converted into the vinylogous Mannich products

⁽⁶⁾ Liu, T.-Y.; Cui, H.-L.; Long, J.; Li, B.-J.; Wu, Y.; Ding, L.-S.; Chen, Y.-C. J. Am. Chem. Soc. 2007, 129, 1878.

⁽⁷⁾ Niess, B.; Jørgensen, K. A. Chem. Commun. 2007, 1620.

⁽⁸⁾ Yamaguchi, A.; Matsunaga, S.; Shibasaki, M. Org. Lett. 2008, 10, 2319.

⁽⁹⁾ Sickert, M.; Schneider, C. Angew. Chem., Int. Ed. 2008, 47, 3631.
(10) (a) Akiyama, T.; Itoh, J.; Yokota, K.; Fuchibe, K. Angew. Chem., Int. Ed. 2004, 43, 1566. (b) Yamanaka, M.; Itoh, J.; Fuchibe, K.; Akiyama, T. J. Am. Chem. Soc. 2007, 129, 6756. (c) Uraguchi, D.; Terada, M. J. Am. Chem. Soc. 2004, 126, 5356. (d) Uraguchi, D.; Sorimachi, K.; Terada, M. J. Am. Chem. Soc. 2005, 127, 9360. See also: (e) Guo, Q.-X.; Liu, H.; Guo, C.; Luo, S.-W.; Gu, Y.; Gong, L.-Z. J. Am. Chem. Soc. 2007, 129, 3790.

⁽¹¹⁾ Reviews: (a) Akiyama, T.; Itoh, J.; Fuchibe, K. Adv. Synth. Catal.
2006, 348, 999. (b) Connon, S. J. Angew. Chem., Int. Ed. 2006, 45, 3909.
(c) Akiyama, T. Chem. Rev. 2007, 107, 5744, and references cited therein.

⁽¹²⁾ Reduction of imines with Hantzsch esters: (a) Rueping, M.; Sugiono, E.; Azap, C.; Theissmann, T.; Bolte, M. Org. Lett. 2005, 7, 3781. (b) Hoffmann, S.; Seayad, A. M.; List, B. Angew. Chem., Int. Ed. 2005, 44, 7424. (c) Storer, R. I.; Carrera, D. E.; Ni, Y.; MacMillan, D. W. C. J. Am. Chem. Soc. 2006, 128, 84. (d) Rueping, M.; Antonchick, A. P.; Theissmann, T. Angew. Chem., Int. Ed. 2006, 45, 3683. (e) Hoffmann, S.; Nicoletti, M.; List, B. J. Am. Chem. Soc. 2006, 128, 13074. (f) Li, G.; Liang, Y.; Antilla, J. C. J. Am. Chem. Soc. 2006, 129, 5830. (g) Rueping, M.; Antonchick, A. P. Angew. Chem., Int. Ed. 2007, 46, 4562. (h) Rueping, M.; Antonchick, A. P. Angew. Chem., Int. Ed. 2008, 47, 5836.

⁽¹³⁾ Åza-Diels-Alder reactions with imines: (a) Itoh, J.; Fuchibe, K.; Akiyama, T. Angew. Chem., Int. Ed. **2006**, 45, 4796. (b) Akiyama, T.; Morita, H.; Fuchibe, K. J. Am. Chem. Soc. **2006**, 128, 13070.

⁽¹⁴⁾ Friedel-Crafts reactions with imines: (a) Uraguchi, D.; Sorimachi, K.; Terada, M. J. Am. Chem. Soc. 2004, 126, 11804. (b) Terada, M.; Sorimachi, K. J. Am. Chem. Soc. 2007, 129, 292. (c) Rowland, G. B.; Rowland, E. B.; Liang, Y.; Perman, J.; Antilla, J. C. Org. Lett. 2007, 9, 2609. (d) Li, G.; Rowland, G. B.; Rowland, E. B.; Antilla, J. C. Org. Lett. 2007, 9, 4065.

⁽¹⁵⁾ For select further applications with imines, see, e.g.: (a) Terada,
M.; Machioka, K.; Sorimachi, K. Angew. Chem., Int. Ed. 2006, 45, 2254.
(b) Rueping, M.; Sugiono, E.; Azap, C. Angew. Chem., Int. Ed. 2006, 45, 2617. (c) Chen, X.-H.; Xu, X.-Y.; Liu, H.; Cun, L.-F.; Gong, L.-Z. J. Am. Chem. Soc. 2006, 128, 14802. (d) Seayad, J.; Seayad, A. M.; List, B. J. Am. Chem. Soc. 2006, 128, 1086. (e) Rueping, M.; Antonchick, A. P.; Brinkmann, C. Angew. Chem., Int. Ed. 2007, 46, 6903.

⁽¹⁶⁾ For select other applications, see, e.g.: (a) Mayer, S.; List, B. Angew. Chem., Int. Ed. 2006, 45, 4193. (b) Rueping, M.; Ieawsuwan, W.; Antonchick, A. P.; Nachtsheim, B. J. Angew. Chem., Int. Ed. 2006, 46, 2097. (c) Rowland, E. B.; Rowland, G. B.; Rivera-Otero, E.; Antilla, J. C. J. Am. Chem. Soc. 2007, 129, 12084. (d) Wang, X.; List, B. Angew. Chem., Int. Ed. 2008, 47, 1119. (e) Cheon, C. H.; Yamamoto, H. J. Am. Chem. Soc. 2008, 130, 9246. (f) Enders, D.; Narine, A. A.; Toulgoat, F.; Bisschops, T. Angew. Chem., Int. Ed. 2008, 47, 5661. (g) Rueping, M.; Theissmann, T.; Kuenkel, A.; Koenigs, R. M. Angew. Chem., Int. Ed. 2008, 47, 6031.

⁽¹⁷⁾ Kang, Q.; Zhao, Z.-A.; You, S.-L. Org. Lett. 2008, 10, 2031.

Table 1. Nucleophile and Catalyst Screen^a



^{*a*} Reactions were performed with imine **1a** (0.20 mmol) and vinylketene silyl *N*,*O*-acetals **5** (3.0 equiv) at -30 °C in a solvent mixture containing equal amounts of 2-propanol, 2-methyl-2-propanol, and 2-methyl-2-butanol with 1.0 equiv of water and 10 mol % of phosphoric acid **4** at 0.1 M concentration. ^{*b*} This reaction was conducted at 0 °C, *E/Z* ratio of 4/1. ^{*c*} Yields refer to chromatographically purified material. ^{*d*} Enantiomeric excesses were determined by HPLC using Chiralcel OD or OD-H columns (see the Supporting Information). The absolute configuration of the products was determined as described in the Supporting Information.

 Table 2. Catalyst Loading^a



^{*a*} Reactions conditions as given below Table 1. ^{*b*} Yields refer to chromatographically purified material. ^{*c*} Enantiomeric excesses were determined by HPLC using Chiralcel OD or OD-H columns (see the Supporting Information).

 Table 3. Scope of the Vinylogous Mukaiyama–Mannich Reaction^a



^{*a*} Reactions conditions as given below Table 1. ^{*b*} Yields refer to chromatographically purified material. ^{*c*} Enantiomeric excesses were determined by HPLC using Chiralcel OD or OD-H columns (see the Supporting Information). The absolute configuration of the products was determined as described in the Supporting Information.

9b—**o** in generally good yields, complete γ -regioselectivity, and up to 92% ee. Aromatic aldimines with ortho-substitutents appear to react slightly less enantioselective than their meta- and para-substituted counterparts. As was already observed for the ester-based silyl dienolates,⁹ 3-thiopheneand 3-furyl-substituted aldimines were among the more selective substrates for the vinylogous Mannich reaction. Aliphatic aldimines decomposed under the reactions conditions and could not be employed as substrates.

We assume that the catalytic cycle of this vinylogous Mannich reaction is similar to our previously discovered process in that water and/or alcohol present in the reaction medium serves to trap the TBS species as silanol or silyl ether, respectively, and reprotonates the phosphate counterion to regenerate the chiral Brønsted acid catalyst.⁹

To demonstrate the practicality of this process, the reaction of imine **1b** was performed at a 10-fold scale (2.0 mmol) with 1 mol % of phosphoric acid **4b** under otherwise identical reaction conditions. Vinylogous Mannich product **9b** was obtained in 91% yield and 88% ee after 5 d at -30 °C (Scheme 2, path a). The reaction may also be performed as a three-component Mannich reaction with aldehyde, amine, and silyl dienolate as starting materials, as was demonstrated for the same reaction (Scheme 2, path b). Mannich product **9b** was now isolated in 96% yield and 86% ee.

Vinylogous Mannich product **9b** was subsequently converted into synthetically useful building blocks with complete retention of configuration (Scheme 3).¹⁸ Treatment of amide

⁽¹⁸⁾ For details see the Supporting Information.

Scheme 2



9b with the Schwartz reagent¹⁹ afforded allylic alcohol **11** in 72% yield, whereas reduction with lithium triethylborohydride²⁰ furnished the saturated alcohol **12** in 73% yield. The latter could be readily converted into piperidine **13** using a Mitsunobu reaction²¹ in good yield.

In conclusion, we have shown that vinylketene silyl *N*,*O*-acetals **5** readily participate in Brønsted acid-catalyzed vinylogous Mukaiyama–Mannich reactions with aromatic and heteroaromatic aldimines and furnish δ -amino- α , β -unsaturated amides **9** in good yields and enantioselectivities. Direct three-component vinylogous Mannich reactions deliver the products with almost identical yield and enantiose-





lectivity, thus avoiding the synthesis of the imines in a separate step. The utility of the vinylogous Mannich products was demonstrated through conversion into various functional building blocks including a short synthesis of the enantiomerically highly enriched 2-phenylpiperidine **13**.

Acknowledgment. We are grateful to Dr. Claudia Birkemeyer and Ramona Oehme (Universität Leipzig) for the MS experiments.

Supporting Information Available: Detailed experimental procedures, characterization, and copies of spectra for the vinylogous Mannich products. This material is available free of charge via the Internet at http://pubs.acs.org.

OL8017374

⁽¹⁹⁾ White, J. M.; Tunoori, A. R.; Georg, G. I. J. Am. Chem. Soc. 2000, 122, 11995.

^{(20) (}a) Brown, H. C.; Kim, S. C. Synthesis 1977, 635. (b) Brown, H. C.;
Kim, S. C.; Krishnamurthy, S. J. Org. Chem. 1980, 45, 1.
(21) (a) Mitsunobu, O.; Yamada, M. Bull. Chem. Soc. Jpn. 1967, 40,

^{(21) (}a) Mitsunobu, O.; Yamada, M. Bull. Chem. Soc. Jpn. 1967, 40, 2380. (b) Mitsunobu, O.; Eguchi, M. Bull. Chem. Soc. Jpn. 1971, 44, 3427.
(c) Mitsunobu, O.; Wade, M.; Sano, T. J. Am. Chem. Soc. 1972, 94, 679.
(d) Mitsunobu, O. Synthesis 1981, 1, 1.